

Fig. 3 Buckling of toroidal segments under external lateral pressure positive gaussian curvature.

Analysis and Results

Stein and McElman⁴ have given a closed-form solution for the buckling of simply supported toroidal segments symmetric about the equator. Specifically, the critical pressure is

$$p_{cr} = k\pi^2 D/r_t l^2$$

where l is the length of the shell, D the flexural stiffness, r_t the throat radius, and k is evaluated by minimizing the expression

$$k = \left[(1 + \beta^2)^4 + (12Z^2/\pi^4)(1 - \alpha\beta^2)^2 \right] / B \tag{1}$$

with respect to β for given Z and α , where $\alpha = r_t/a$, $\beta = nl/(\pi r_t)$, a is the radius of curvature of the meridian, and

$$Z = (l^2/r_t t)(1-\mu^2)^{1/2}$$

$$B = \beta^2 (1+\beta^2)^2 \quad \text{for lateral pressure}$$

 $B = 0.5(1 + 2\beta^2 - \alpha\beta^2)(1 + \beta^2)^2$ for hydrostatic pressure

Abel and Billington³ suggest that Eq. (1) be applied to cooling towers by considering only the top part of the tower for which the throat is the midheight. As a check of the applicability of this procedure, the finite-element method was used to analyze the toroidal segment shown in Fig. 1.

Conclusions

The Stein and McElman⁴ theory is limited to shallow shell approximations and to buckling near the equator of a toroidal shell segment. For the lateral pressure case and particular shell geometry, it was found that a more accurate finite-element shell buckling solution agree approximately with the Stein and McElman⁴ predictions only for the positive curvature shell. Experimental results from Veronda⁵ were in good agreement with the finite-element solution and were far away from predicted values obtained from the Stein and McElman⁴ design curves. It appears that the assumptions used by Stein and McElman are very restrictive and must be checked for each specific design case by using a more accurate shell stability solution.

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CO, Laser Radiation Absorption in SF₆-Air Mixtures

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I. Introduction

¬ ASEOUS sulfur hexafluoride (SF₆) is a strong absorber of Tradiation at 10.6μ . This wavelength also corresponds to the output from common CO2 lasers. As a result of this resonance phenomena, SF₆ has found applications in Q switching and mode locking CO₂ gas lasers, and has been useful in other aspects of laser physics as itemized by Steinfeld et al.1 In turn, these applications have prompted fundamental measurements of SF₆ infrared absorption coefficients, 2-8 and encouraged studies of its saturation and self-induced transparency effects.5

The addition of SF_6 to air, and the subsequent laser radiation absorption in such SF_6 -air mixtures, has potential applications in aerodynamics for tailoring flows for useful purposes. Several examples are discussed in Ref. 10. Moreover, it has potential as a unique energy conversion mechanism, such as in photon engines.¹¹ The purpose of this Note is to present physical data on the 10.6μ radiation absorption and saturation properties of SF s-air mixtures; these data are needed for future aerodynamic applications.

II. Absorption Coefficient

The molecular properties of SF₆ are summarized in Refs. 10 and 12. The absorption of CO₂ laser radiation by SF₆ occurs, in part, between the ground state and the first excited level of the v_3 vibrational mode. Hot band absorption also occurs from the first excited level of the v_6 mode to the first combined $(v_3 + v_6)$ level. Moreover, the present results, as well as those of Ref. 13, imply that additional hot band absorption from other low-lying vibrational levels must be considered. For each of these vibrational transitions, the 10.6μ laser radiation is absorbed by a large number of overlapping vibration-rotation lines; hence, the pressure broadened absorption coefficient for each vibrational transition is 14

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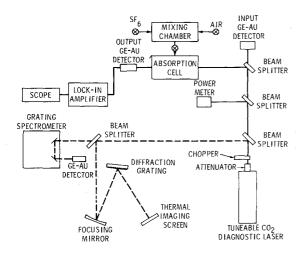


Fig. 1 Schematic diagram of absorption cell apparatus.

$$\alpha_{0}(v) = \frac{\lambda^{2}}{4\pi^{2} \Delta v} \sum_{J} A_{21J} \left(N_{1J} - \frac{g_{1}}{g_{2}} N_{2J\pm 1} \right) \left[4 \left(\frac{v - v_{J}}{\Delta v} \right)^{2} + 1 \right]^{-1}$$

where $N_{2J\pm 1}$ and N_{1J} are the populations of the upper and lower vibration-rotation energy levels, respectively, and λ , Δv , A_{21J} , g_1 , and g_2 are the wavelength, line width, Einstein coefficient, and statistical weights, respectively. Replacing the sum by an integral, the absorption coefficient for CO_2 laser radiation due to a given vibrational transition in SF_6 becomes

$$\alpha_0 = A_{21} \phi_1 [N_1 - (g_1/g_2)N_2]$$

where ϕ_1 is a function of temperature and all other subscripts refer to the respective vibrational levels. From this equation, the effect of a large number of overlapping lines is to cause α_0 to vary linearly with pressure. This is in contrast to the case of a single pressure broadened line, where α_0 at line center is independent of pressure.

The present results extend previous measurements²⁻⁸ of the 10.6μ absorption coefficient for pure SF_6 to higher pressures and temperatures. Moreover, the present measurements give the first results for α_0 in SF_6 -air mixtures of interest in aerodynamic applications. A schematic of the experimental setup for the present absorption coefficient measurements is shown in Fig. 1.

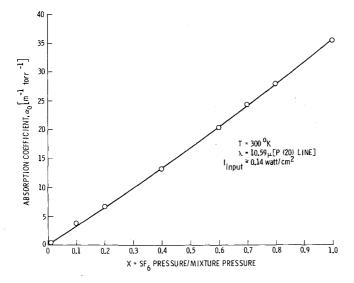


Fig. 2 Variation of absorption coefficient with SF_6 mole fraction in SF_6 -air mixtures.

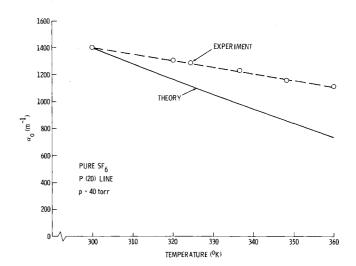


Fig. 3 Temperature variation of SF₆ absorption coefficient.

A portion of the chopped output beam from a stable tuneable CO₂ laser passes through a 5-mm-long absorption cell, which contains a regulated SF₆-air mixture at desired pressure and temperature. Five basic laser measurements are monitored: input intensity to the absorption cell, output intensity from the absorption cell, average power, wavelength, and mode shape. The apparatus and experimental procedure are described in Ref. 10. The data obtained for a large range of mixture pressure and mole fractions, as well as for CO_2 laser wavelengths from the P(10) to P(32) transitions, are also given in Ref. 10. These data show that α_0 is indeed a linear function of pressure for all wavelengths and mole fractions considered. Some results are shown in Fig. 2, where the absorption coefficient in units of (m⁻¹ torr⁻¹) is given as a function of the mole fraction of SF₆. The linear variation shown in this figure clearly demonstrates that α_0 is dependent on the number density of SF₆, and is independent of the number density of air. Therefore, the pressure broadening cross sections for SF₆-air collisions must be of the same order as those for $SF_6 - SF_6$ collisions. Moreover, for pure SF₆, the present data indicate that $\alpha_0 = 35.4 \text{ m}^{-1} \text{ torr}^{-1}$; this agrees very well with the value of 35.7 m⁻¹ torr⁻¹ obtained from the data of Ref. 8.

The variation of α_0 over a moderate range of T is shown in Fig. 3. The observed reduction of α_0 with increasing T is not as large as expected on theoretical grounds including just the ground state and the first hot band absorptions.¹⁰ This implies that additional hot band absorptions may contribute to α_0 , even at room temperature.

III. Saturation Effects

If the laser beam is intense enough and the gas density is low enough, molecular excitation due to photon absorption can dominate the de-excitation due to molecular collisions, causing the ground state population to be depleted and the excited levels to become overpopulated. This vibrational nonequilibrium causes α to become a function of the incident laser intensity, $I_{\rm in}$, i.e., α is influenced by saturation effects.

Saturation intensities in pure SF_6 have been measured previously at low pressures. ^{6,7} However, no results (experimental or theoretical) at near-atmospheric pressures in SF_6 -air mixtures of aerodynamic interest have been available. Such results are given in Fig. 4; here, α is given as a function of $I_{\rm in}$ for a 10% SF_6 -90% air mixture at pressures corresponding to standard altitudes of sea level, 10,000 ft and 20,000 ft. These are theoretical results obtained from the kinetic model and nonequilibrium analysis described in Ref. 10. A constant temperature gas is assumed, hence isolating the purely nonequilibrium effects from

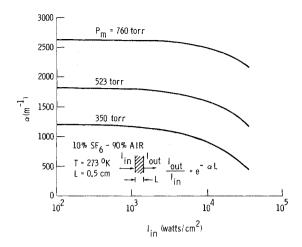


Fig. 4 Predicted saturation effects in a constant temperature SF₆-air mixture at near-atmospheric pressures.

the influence of bulk heating. 10 The vibrational relaxation times used in this analysis are from Ref. 1, where $p\tau_{SF_6-SF_6} = 122$ μ sec-torr, and $p\tau_{SF_6^-N_2} = 103 \mu$ sec-torr. This latter value is adopted for $p\tau_{SF_6-air}$. The results shown in Fig. 4 suggest that saturation effects for a typical SF₆-air mixture at nearatmospheric pressure are not important until Iin exceeds 104 w/cm².

IV. Conclusions

From the results given in this Note, and from the extensive data shown in Refs. 10 and 12 (on which this Note is based), the following conclusions are made.

- 1) Pure SF₆ is a powerful absorber of 10.6μ radiation. For example, the small-signal absorption coefficient, α_0 for SF₆ is on the order of 3540 m⁻¹ at p = 100 torr.
- 2) SF₆-air mixtures are also strong absorbers of 10.6μ radiation. For example, $\alpha_0 = 2740 \text{ m}^{-1}$ in a 10% SF₆-90% air mixture at 1 atm.
- 3) The experimental variation of α_0 with p is linear. This is consistent with the theoretical prediction for absorption due to a large number of closely spaced overlapping lines.
- 4) Comparison between experimental data and theory imply that additional (and as yet unidentified) hot band transitions are contributing to SF₆ absorption at room temperature and above.
- 5) At pressures near atmospheric, the onset of saturation for a 10% SF₆-90% air mixture occurs above intensities of 10^4 w/cm².

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Reattachment of a Separated Boundary Layer to a Convex Surface

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RAPERIMENTS on simplified models of the interaction between a transverse two-dimensional jet and an enveloping supersonic flow have been conducted at the Gas Dynamics Labs. of The Univ. of Michigan. One of the simplified models consisted of a flat plate upon which was mounted a solid obstacle representing the jet (see Fig. 1). The shape chosen for this obstacle was a cylinder of semicircular cross section, the flat side of which was placed against the flat plate, with the cylinder axis perpendicular to the flow direction. The forward part of this circular cylinder thus represents a portion of the dividing stream surface between an imagined two-dimensional transverse jet and the main airstream. The flow ahead of this simulated jet behaves very much like that ahead of a real jet, in that it separates from the plate at some distance ahead of the jet, and then reattaches to the convex (jet) surface while undergoing a strong interaction shock. The flow pattern is also quite similar to that ahead of a forward-facing step, with the important difference that reattachment is not constrained to occur at a fixed point (the corner of the step). This Note describes experimental determinations of the location of reattachment on the curved surface, and correlation with a simple theory.

The tests were conducted at Mach number 3.9 and Reynolds number 120,000 per inch in the 8- by 13-in. supersonic wind tunnel of The Univ. of Michigan, on the two-dimensional model shown in cross section in Fig. 1. The semicircular cylinder had a diameter of 1 in. and a span of 3 in. The flat plate on which it

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